## Additives

### Most chemicals in our food are there naturally

- Intentional Additives= additives or ingredients added directly and intentionally for their beneficial effects salt, sugar, processing aids, preservatives, vitamin and mineral supplements
- Unintentional Additives = get into food "by mistake". Microbes and toxins, lubricants, detergents, drug/pesticide residues, environmental contaminants.

#### Exhibit 11-1 Toxic Chemicals in Foods

#### NATURAL

- normal components of natural food products
- natural contaminants of natural food products
  - -microbiological origin: toxins
  - nonmicrobiological origin: toxicants
    (e.g., Hg, Se) consumed in feeds by animals used as food sources

#### MAN-MADE

 agricultural chemicals (e.g., pesticides, fertilizers)

- · food additives
- chemicals derived from food packaging materials
- chemicals produced in processing of foods (e.g., by heat, ionizing radiation, smoking)
- · inadvertent or accidental contaminants
  - -food preparation accidents or mistakes
  - -contamination from food utensils
  - -environmental pollution
  - -contamination during storage or transport

## Risky "Additives" in decreasing order by risk

- Pathogenic microbes (bacteria, viruses, parasites)
- Naturally occurring toxins (aflatoxin, marine biotoxins, etc)
- Naturally occurring anti-nutritional factors
- Allergens
- Pesticide residues and their metabolites
- Drug residues and their metabolites
- Environmental contaminants (from pollution)
- Intentional additives (used at higher than approved levels)

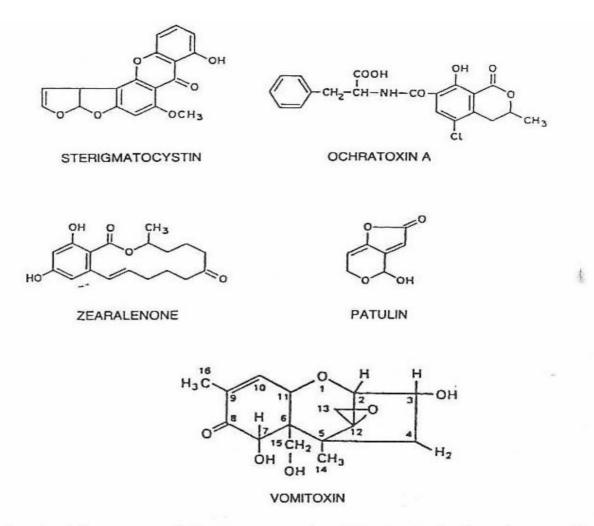


Figure 11–17 Chemical Structures of Sterigmatocystin, Ochratoxin A, Zearalenone, Deoxynivalenol (Vomitoxin), and Patulin. *Source:* From P.M. Scott, The Analysis of Foods for Aflatoxins and Other Fungal Toxins: A Review, *Can. Inst. Food Technol. J.*, Vol. 2, pp. 173–177, 1969; P.M. Scott et al., Effects of Experimental Flour Milling and Breadbaking on Retention of Deoxynivalenol (Vomitoxin) in Hard Red Spring Wheat, *Cereal Chem.*, Vol. 60, pp. 421–424, 1983.

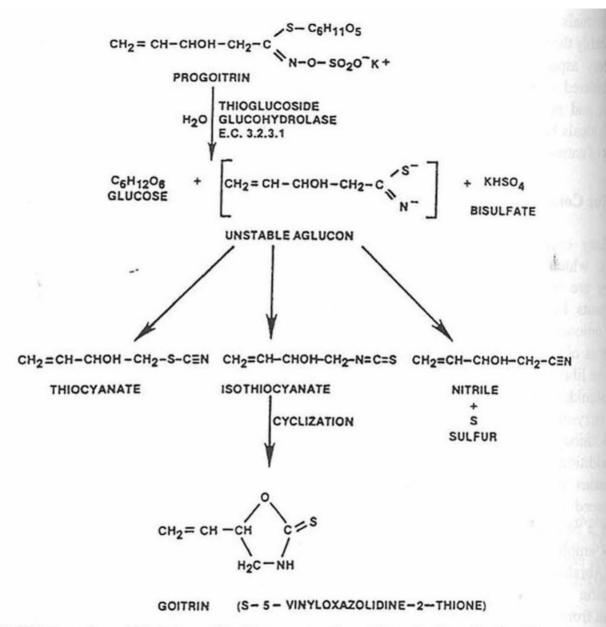


Figure 11–18 Formation of Goitrin and Isothiocyanates from Glucosinolates in Cruciferous Products. Source: From D.K. Salunkhe and M.T. Wu, Toxicants in Plants and Plant Products, Food Sci. Nutr., Vol. 9, pp. 265–324, 1977.

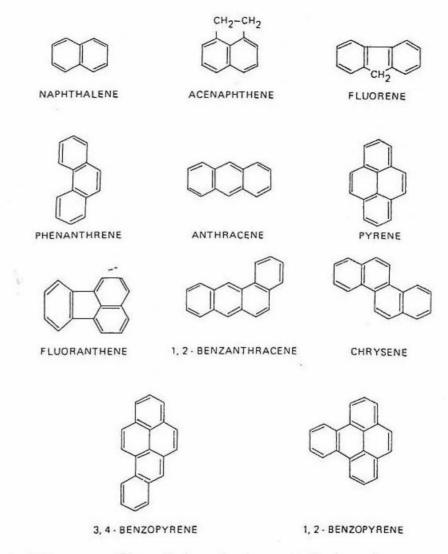


Figure 11–15 Chemical Structure of Some Polycyclic Aromatic Hydrocarbons

# Aspartame – 7.5-13 mg/Kg/d presume 70 Kg person

- On can soda- 190 mg or 2.7 mg/Kg/day
- 5 cans per day 13.5 mg

# Categories and Functions of Intentional Ingredients

- 1. Add bioactive compounds
- 2. Maintain palatability and wholesomeness
- 3. Enhance appeal-sensory characteristics
- 4. Leavening agents
- 5. pH control
- 6. Processing aids

## Add bioactive compounds

- Vitamins and minerals
- Bioactive agents (antioxidants, plant sterols)
- Stimulants

Figure 11-19 The Structure of Caffeine 1,3,7- Trimethylxanthine

## 2. Maintain palatability and wholesomeness

- *Antimicrobials* nisin, benzoates, sorbate, salt
- Antioxidants –BHA, BHT, vitamin E, Vitamin C
- Antibrowning agents -sulfite, EDTA, acidulants

Table 11-1 Applications of Sorbates as Antimicrobial Agents

Products	Levels (%)
Dairy products: aged cheeses, processed cheeses, cottage cheese, cheese spreads, cheese dips, sour cream, yogurt	0.05-0.30
Bakery products: cakes, cake mixes, pies, fillings, mixes, icings, fudges, toppings, doughnuts	0.03-0.30
Vegetable products: fermented vegetables, pickles, olives, relishes, fresh salads	0.02-0.20
Fruit products: dried fruit, jams, jellies, juices, fruit salads, syrups, purees, concentrates	0.02-0.25
Beverages: still wines, carbonated and noncarbonated beverages, fruit drinks, low-calorie drinks	0.02-0.10
Food emulsions: mayonnaise, margarine, salad dressings	0.05-0.10
Meat and fish products: smoked and salted fish, dry sausages	0.05-0.30
Miscellaneous: dry sausage casings, semimoist pet foods, confectionery	0.05-0.30

Source: Reprinted with permission from J.N. Sofos and F.F. Busta, Sorbic Acid and Sorbates, in *Antimicrobials in Foods*, P.M. Davidson and A.L. Branen, eds., p. 62, 1993, by courtesy of Marcel Dekker, Inc.

#### 440 Principles of Food Chemistry

Figure 11-3 Chemical Structure of the Active Antioxidant Principles in Rosemary

#### 3. Enhance appeal-sensory characteristics

- Flavors –spices, natural extracts, artificial flavors
- Flavor enhancers MSG, protein hydrolysates, yeast autolysate, nucleic acids
- Sweetners sugar, artificial sweetners, sugar alcohols
- Colorants natural and artificial
- Texturizing agents emulsifiers (lecithin, egg yolk), stabilizers (gums, starch),
- Water binding agents humectants like sorbitol, sugar, milk powder
- *Thickeners* gums

#### 444 Principles of Food Chemistry

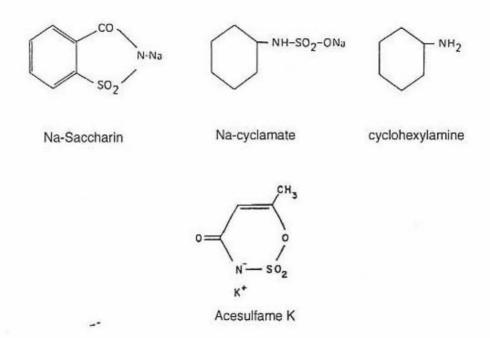


Figure 11-4 Chemical Structure of Sodium Saccharin, Sodium Cyclamate, Cyclohexylamine, and Acesulfame K

Table 11-5 Major Categories of Natural Food Colors and Their Sources

Colorant	Sources
Anthocyanins	Grape skins, elderberries
Betalains	Red beets, chard, cactus fruits, pokeberries, bougainvillea, amaranthus
Caramel	Modified sugar
Carotenoids	
Annatto (bixin)	Seeds of Bixa orellana
Canthaxanthin	Mushrooms, crustaceans, fish, seaweed
β-apocarotenal	Oranges, green vegetables
Chlorophylls	Green vegetables
Riboflavin	Milk
Others	
Carmine (cochineal extract)	Coccus cati insect
Turmeric (curcuma)	Curcuma longa
Crocetin, crocin	Saffron

Source: Reprinted with permission from R.L. Newsome, Natural and Synthetic Coloring Agents, in Food Additives, A.L. Branen, P.M. Davidson, and S. Salminen, eds., p. 333, 1990, by courtesy of Marcel Dekker, Inc.

Exhibit 11-2 Color Additives Permitted for Food Use in the United States and Their Common Names

- · FD&C red no. 3 (erythrosine)
- FD&C red no. 40 (allura red)
- · FD&C orange B
- FD&C yellow no. 6 (sunset yellow)
- FD&C yellow no. 5 (tartrazine)
- FD&C green no. 3 (fast green)
- FD&C blue no. 1 (brillian blue)
- FD&C blue no. 2 (indigotine)
- · Citrus red no. 2

Source: Reprinted with permission from R.L. Newsome, Natural and Synthetic Coloring Agents, in *Food Additives*, A.L. Branen, P.M. Davidson, and S. Salminen, eds., p. 344, 1990, by courtesy of Marcel Dekker, Inc.

## 4. Leavening agents

- Yeast
- Baking powder

#### 5. pH control

- Acids citric, lactic, malic, lemon juice, phosphoric
- Bases phosphate salts

Table 11-4 Properties of Some Common Food Acids

Property	Acetic Acid	Adipic Acid	Citric Acid	Fumaric Acid	Glucono- Delta- Lactone	Lactic Acid	Malic Acid	Phosphoric Acid	Tartaric Acid
Structure	СН₃СООН	соон   сн₂	COOH   CH₂	нооссн    нссоон	о=с	СН <sub>3</sub>   Н—С—ОН	соон		соон   н—с—он
		 CH₂ 	но-с-соон		носн о	СООН	CH <sub>2</sub>		но-с-н
		CH₂       CH₂ 	СН <sub>2</sub>   СООН		нс		СООН		соон
		ĊООН			ĊH <sub>2</sub> OH			479	
Empirical formula	$C_2H_4O_2$	$C_6H_{10}O_4$	$C_6H_8O_7$	$C_4H_4O_4$	$C_6H_{10}O_6$	$C_3H_6O_3$	$C_4H_6O_5$	H <sub>3</sub> PO <sub>4</sub>	$C_4H_6O_6$
Physical form	Oily Liquid	Crystalline	Ċrystalline	Crystalline	Crystalline	85% Water Solution	Crystalline	85% Water Solution	Crystalline
Molecular weight	60.05	146.14	192.12	116.07	178.14	90.08	134.09	82.00	150.09
Equivalent weight	60.05	73.07	64.04	58.04	178.14	90.08	67.05	27.33	75.05
Sol. in water (g/100 mL solv.)	∞	1.4	181.00	0.63	59.0	00	144.0	∞	147.0
Ionization constants									
K <sub>1</sub>	8 × 10 <sup>-5</sup>	$3.7 \times 10^{-5}$	$8.2 \times 10^{-4}$	1 × 10 <sup>-3</sup>	2.5 × 10 <sup>-4</sup> (gluconic acid)	1.37 × 10 <sup>-4</sup>	4 × 10 <sup>-4</sup>	$7.52 \times 10^{-3}$	$1.04 \times 10^{-3}$
K <sub>2</sub>		$2.4 \times 10^{-6}$	$1.77 \times 10^{-5}$	$3 \times 10^{-5}$			$9 \times 10^{-6}$	$6.23 \times 10^{-8}$	$5.55 \times 10^{-5}$
$K_3$			$3.9 \times 10^{-6}$				3., 10	$3 \times 10^{-13}$	0.55 × 10

### 6. Processing aids

- Moisture control phosphate salts,
- Filtration aids diatomaceous earth
- Anticaking silica gel, Ca phosphate
- Enzymes filtration aid or clarification
- Emulsifiers and defoamers (lecithin, oils)